

# Photocatalytic degradation of cobalt cyanocomplexes in a novel LED photoreactor using TiO<sub>2</sub> supported on borosilicate sheets: A new perspective for mining wastewater treatment

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## Abstract

The photocatalytic degradation of hexacyanocobaltate ion ( $[\text{Co}(\text{CN})_6]^{3-}$ ) by TiO<sub>2</sub> supported on borosilicate sheets was studied in a novel photoreactor operated under UVA-LEDs irradiation. The presence and absence of O<sub>2</sub> during the process, and the reuse of the TiO<sub>2</sub>-impregnated sheets were evaluated. The semiconductor was supported by the dip-coating method, and the influence of g TiO<sub>2</sub>/g Methanol (MetOH) ratio in the suspension and substrate roughness modification were evaluated in the detachment percentage and adherence of TiO<sub>2</sub>. The highest semiconductor detachment percentage was obtained at 0.36 g TiO<sub>2</sub>/g MetOH ratio. Besides, it was determined that for the ten layers supported, the substrate roughness modification does not influence the detachment of TiO<sub>2</sub>. In the first photocatalytic tests, the results show that a better degradation performance was achieved by direct photolysis than photocatalysis according to the higher concentration of CN<sup>-</sup> released from the cyanocomplex. However, a decrease in the concentration of cobalt in solution was not observed. When the sheets were reused, a decrease of 10% in the concentration of cobalt was achieved, and 14% of CN<sup>-</sup> was released from the cyanocomplex. This was attributed to the formation of microchannels, hollows amongst other imperfections that increase the surface area and active sites of the coating when TiO<sub>2</sub> peels off. The simplified kinetics analysis shows that for UV + O<sub>2</sub> (oxic environment) UV + N<sub>2</sub> (anoxic environment) similar kinetic parameters were obtained, indicating that both processes follow the same homogeneous pathway in the degradation of  $[\text{Co}(\text{CN})_6]^{3-}$ . However, for the UV + TiO<sub>2</sub>+O<sub>2</sub> and UV + TiO<sub>2</sub>+N<sub>2</sub>, their different kinetic parameters suggest a non-homogeneous degradation mechanism with different pathways induced by the presence or absence of O<sub>2</sub>.

## Keywords

Dissolved oxygen, UVA-LEDs, Detachment percentage, Roughness modification, Photocatalyst reuse, Heavy metals